

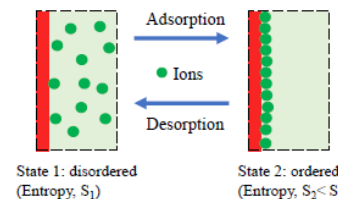
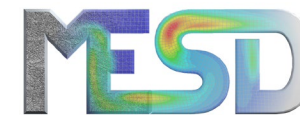
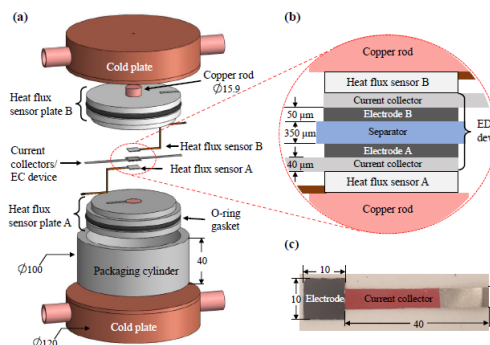
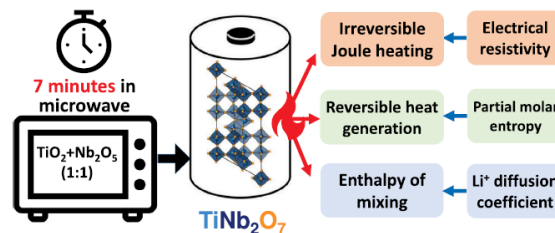
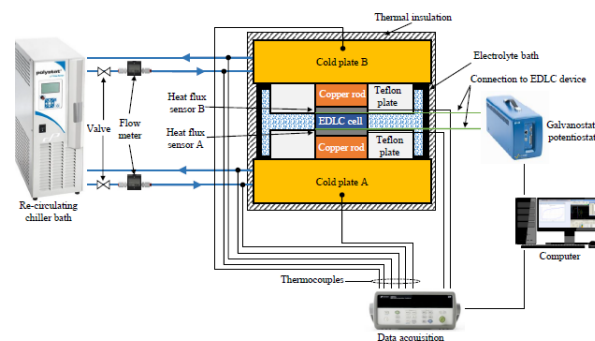


Dr. Ampol Likitchachawankun

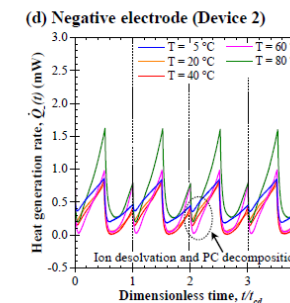
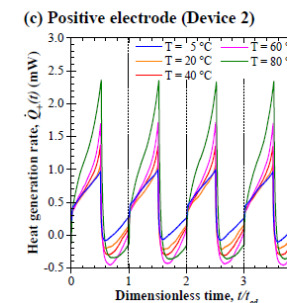
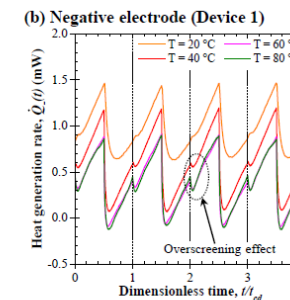
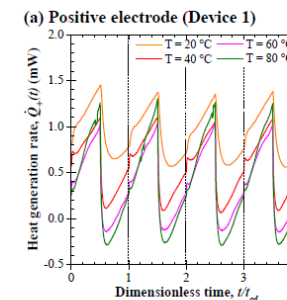
ดร. อัมพล ลิขิตชัชวาลกุล

Research Area:

- Heat and Mass Transfer
- Interfacial and Transport Phenomena
- Renewable Energy Engineering
- Thermal Power Plants
- Energy Storage Devices
- Microelectromechanical Systems (MEMS)



$\delta Q = TdS$
 $Q_{rev} = TdS/dt$
 Adsorption: State 1 \rightarrow State 2, where $S_2 < S_1$.
 $dS/dt < 0$ therefore $Q_{rev} < 0$.
 Heat was released from the system (exothermic).
 Desorption: State 2 \rightarrow State 1, where $S_2 < S_1$.
 $dS/dt > 0$ therefore $Q_{rev} > 0$.
 Heat was absorbed into the system (endothermic).



Collaboration:



Mechanical Engineering Simulation and Design

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